編號: 148

## 國立成功大學 109 學年度碩士班招生考試試題

系 所:環境工程學系

考試科目:環境化學及環境微生物學

考試日期:0210,節次:2

第1頁,共2頁

※ 考生請注意:本試題可使用計算機。 請於答案卷(卡)作答,於本試題紙上作答者,不予計分。

1. Water becomes scarcer due to climate change. Many countries are failing to satisfy the basic need to provide sufficient quantities of water of acceptable quality. Reclaim water can provide a reliable water source for industry, irrigation, and municipal use.

The treated effluent of a chemical plant contains the following ions: Na<sup>+</sup> 0.02M; Mg<sup>2+</sup> 0.015 M; Ca<sup>2+</sup> 0.01M; K<sup>+</sup> 0.001M; Cl<sup>-</sup> 0.002M; HCO<sub>3</sub><sup>-</sup> 0.001M; NO<sub>3</sub><sup>-</sup> 0.002M; and SO<sub>4</sub><sup>2-</sup> 0.012M

- (a) Please estimate the pH value of this effluent. Please state your reason clearly for full credits. (10%)
- (b) Please determine the osmotic pressure difference across a semipermeable membrane that had this treated effluent on one side and freshwater water (i.e. ion-free water) on the other. The temperature is 25 ° C. (10%)
- (c) If you want to obtain 80% of the freshwater production from this effluent, what minimum pressure would be required to balance the osmotic pressure difference that will occur? (5%)
- 2. Incineration of municipal solid waste almost always implies the burning of some polyvinyl chloride (PVC), a common form of plastic found in construction and many household goods, including packaging and toys. PVC is a polymeric chain repeating a monomer.
  - (a) Please write the chemical structure of this monomer of PVC and the reaction describing the high-temperature combustion of the PVC monomer. (7%)
  - (b) Calculate the volume of air at T = 20°C and P = 1 atm required to incinerate 2 kg of PVC. Express your answer in liters. (8%)
- 3. A municipal wastewater treatment plant is processing a waste flow with a 5-day BOD of 200 mg/L at 20° C. If the BOD rate constant k at 20°C is 0.23 day<sup>-1</sup>, please calculate the ultimate BOD (mg/L) of the raw wastewater at 20°C. (10%)

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考試科目:環境化學及環境微生物學

第2頁,共2頁

考試日期:0210,節次:2

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## ■ 環境微生物學考題

- 4生活污水中含有蛋白質、氨等物質,如果沒有經過處理就排入承受水體,容易造成水質污染。氮元素有不同價數的氧化態可與其他元素結合,例如胺基酸、氨、硝酸鹽等,環境中的微生物可以代謝這些含氮元素的化合物,使得氮元素不斷地在生態系統中循環,稱為氮循環(Nitrogen cycle)。 請繪製出氮循環 (10分),並條列說明構成氮循環的反應 (14分)。在環境工程領域,常常應用氮循環反應去除污水中含氮化合物,請試列舉一種生物處理方法,並說明如何轉換水中氨氮變成穩定無害的物質,達到去除污染物的目的(5分),以及參與其中的各種氮轉換反應的電子供給者、最終電子接受者(6分),以及一種代表微生物的屬名(genus name) (9分)。
- 5、請試說明 Woese 與 Fox (1977)提出的微生物分類系統以及所根據的基因分子 (6分)。