系所組別: 高分子工程系碩士班乙組

科 目: 有機化學

## 乙組:有機化學;共12大題,總分100分;請於答案卷內依序作答。

- 1. What configuration of the products would you expect from SN<sub>2</sub> reaction of the optically active (R)-2-octanol with following sets of reagents? (1) (i) PBr<sub>3</sub>/ether,
  - (ii) CH<sub>3</sub>CH<sub>2</sub>O'Na<sup>+</sup>; (2) (i) p-TosCl\*/pyridine, (ii) CH<sub>3</sub>CH<sub>2</sub>O'Na<sup>+</sup>.

(\*: p-toluenesulfonyl chloride) (6%)

2. Identify the reagents a-j in the following scheme: (10%)

$$CH_2OH \xrightarrow{c} O \xrightarrow{b} OH \xrightarrow{c} OH$$

$$CH_2OH \xrightarrow{c} CHO \xrightarrow{f} QH$$

$$\downarrow i$$

$$\downarrow g$$

$$NH=C \xrightarrow{j} NH_2$$

- 3. Write chemical equations of the following reactions: (you should write all steps of the reaction and any other chemical or reaction condition necessary) (15%)
  - (1) Sandmeyer reaction
  - (2) Aldol condensation reaction
  - (3) Michael reaction
  - (4) Fischer esterification reaction
  - (5) Claisen condensation reaction
- 4. Synthesize the following compounds: (In addition to the starting materials, you can use any chemicals needed) (9%)
  - (1) from C<sub>6</sub>H<sub>6</sub> to m-Cl-C<sub>6</sub>H<sub>4</sub>COOH
  - (2) from CH<sub>3</sub>CH<sub>2</sub>OH to CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CHO
  - (3) from (CH<sub>3</sub>)<sub>3</sub>C-OH to (CH<sub>3</sub>)<sub>3</sub>C-O-CH<sub>2</sub>CH<sub>3</sub>

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5. Name the following compounds: (6%)

6. Refer to the following equation to answer the questions below. (6%)

$$pKa = 18$$

$$(CH_3)_3C - \ddot{O} - H + K^{\oplus \odot}\dot{O}H \longrightarrow (CH_3)_3C - \ddot{O} \cdot K + H_2O$$
A. B. C. D.

- (6a). Which is the strongest Brønsted-Lowry acid in the equation?
- (6b). Which is the strongest Brønsted-Lowry base in the equation?
- (6c). Will this reaction take place as written? Please explain.

7. Below are the two chair conformations of a 1,2,4-trimethylcyclohexane. Estimate the amount of 1,3-diaxial strain in each conformer and predict which conformer is most stable by calculating the energy associated with their conformation. (giving that 1,3-diaxial strain: H-CH<sub>3</sub>: 3.8 KJ/mol, CH<sub>3</sub> gauche interaction: 3.8 kJ/mol) (6%)

$$CH_3$$
 $CH_3$ 
 $CH_3$ 

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- 8. Choose the best reagent from the list below for carrying out each transformation.(12%)
  - - 2. Zn, H<sub>3</sub>O<sup>+</sup>
  - b. 1. BH<sub>3</sub>, THF
    - - 2. H<sub>2</sub>O<sub>2</sub>, NaOH, H<sub>2</sub>O
  - c. CHCl<sub>3</sub>, KOH
  - d. H<sub>2</sub>O, H<sub>2</sub>SO<sub>4</sub>, heat
- g. CH<sub>2</sub>I<sub>2</sub>, Zn(Cu)

f. KMnO<sub>4</sub>, acid

1. OsO<sub>4</sub>

h. 1. Hg(OAc)<sub>2</sub>, H<sub>2</sub>O

2. NaHSO<sub>3</sub>, H<sub>2</sub>O

2. NaBH<sub>4</sub>

8a.

8b.

CH<sub>3</sub>

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9. Reaction of HBr with 3-methylcyclohexene yields a mixture of four products: cisand trans-1-bromo-3-methylcyclohexane and cis- and trans-1-bromo-2-methylcyclohexane. However, the analogous reaction of HBr with 3-bromocyclohexene yields trans-1,2-dibromocyclohexane as the sole product. Draw structures of the possible intermediates, and then explain why only a single product is formed in the reaction of HBr with 3-bromocyclohexene. (4%)

10. One of the following molecules (a)~(d) is D-erythrose 4-phosphate, which has R stereochemistry at both chirality centers. (6a). Which of the structures is it? (6b). Which of the remaining structures is the enantiomer of D-erythrose 4-phosphate? (6c). and which are diastereomers? (6%)

(a) 
$$H = C = O$$
 (b)  $H = C = O$  (c)  $H = C = O$  (d)  $H = C = O$  H  $= C = O$  CH<sub>2</sub>OPO<sub>3</sub><sup>2-</sup> CH<sub>2</sub>OPO<sub>3</sub><sup>2-</sup> CH<sub>2</sub>OPO<sub>3</sub><sup>2-</sup>

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11. Propose structures for compounds that fit the following descriptions: (12 %)

a)  $C_{10}H_{14}$ 

<sup>1</sup>H NMR:  $7.18 \delta$  (4 H, broad singlet);

2.70  $\delta$  (4 H, quartet, J = 7 Hz);

1.20  $\delta$  (6 H, triplet, J = 7 Hz)

IR: 745cm<sup>-1</sup>

 $(b)C_{10}H_{14}$ 

<sup>1</sup>H NMR: 7.0  $\delta$  (4 H, broad singlet); 2.85  $\delta$  (1 H, septet, J = 8 Hz);

2.28  $\delta$  (3 H, singlet); 1.20  $\delta$  (6 H, doublet, J = 8 Hz)

IR: 825cm<sup>-1</sup>

12. Propose structures for aromatic hydrocarbons that meet the following descriptions: (8%)

- a) C<sub>10</sub>H<sub>14</sub>; gives only one C<sub>10</sub>H<sub>13</sub>Cl product on substitution with chlorine
- b) C<sub>8</sub>H<sub>10</sub>; gives three C<sub>8</sub>H<sub>9</sub>Br products on substitution with bromine