1. Give the IUPAC name for each of the following compound: (15%)

- 2. Draw the structure for each of the following compound. (15%)
 - (a) Isobutylcyclopentane
 - (b) trans-2-Methylcyclopentanol
 - (c) 3-Hydroxybutanal
 - (d) Acetone cyanohydrin
 - (e) Bisphenol A
- 3. For the all C4 alkanes: butane, isobutane, and cyclobutane, order the (a) boiling point; and (b) heat of combustion? (Ex. butane > isobutene > cyclobutane) (6%)
- 4. For the primary, secondary, and tertiary carbons, order (a) the stability of the carbocation; and (b) the bond dissociation energy (BDE) of C-H bond. (6%)
- 5. Order the acidity of the following compounds: (a) ethane, ethylene, and acetylene; (b) benzoic acid, phenol, and o-nitrophenol. (6%)
- 6. Write the structure of the (a) conjugated acid; and (b) conjugated base of methanol. (6%)

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7. Draw the structure of the main organic product for each of the following reaction. (24%)

(a)
$$OH \xrightarrow{H_2SO_4} \xrightarrow{Br_2} \xrightarrow{H_2O}$$
 (e) HBr acetic acido

(b) Br_2 HBr peroxide

(c) Br OK OK OK OEf

- 8. Draw the structures of the monomer and polymer of (a) poly(methyl methacrylate); and (b) poly(lactic acid). (10%)
- 9. Write the reactions of the initiation, propagating, and termination steps for the free-radical polymerization of ethylene. (12%)