

系所組別： 化學工程學系乙組

考試科目： 無機化學及分析化學

考試日期：0219 · 節次：2

※ 考生請注意：本試題 可 不可 使用計算機**Inorganic Chemistry and Analytical Chemistry (total points: 100)**

Part I: Inorganic Chemistry (total points: 50)

- (1) Determine the possible microstates for an  $s^1 p^1$  configuration, and use them to prepare a microstate table. (10 pts)
- (2) Determine the possible values of  $J$  (total angular momentum quantum number) for the carbon terms. (5 pts)
- (3) Confirm that a (real)  $2p_x$  orbital is orthogonal to a (real)  $2p_y$  orbital of the same atom. (5 pts)
- (4) Reduce the following representations to their irreducible representations in the point group indicated: (10 pts)

$C_{2h}$	$E$	$C_2$	$i$	$\sigma_h$
$\Gamma$	4	0	2	2

- (5) Using the angular overlap model, determine the splitting pattern of  $d$  orbitals for a tetrahedral complex of formula  $MX_4$ , where  $X$  is a ligand that can be act as  $\sigma$  donor and  $\pi$  donor. (10 pts)
- (6) For which  $d^n$  configurations would no Jahn-Teller splitting expected for the tetrahedral case (ignore possible low-spin case) (10 pts)

(背面仍有題目,請繼續作答)

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## Part II: Analytical Chemistry (Total points: 50)

- (1) A spectrum has a signal-to-noise ratio of 8/1. How many spectra must be averaged to increase the signal-to-noise ratio to 20/1? (5 pts)
- (2) A solution was prepared by mixing 10.00 mL of unknown (X) with 5.00 mL of standard (S) containing 8.24  $\mu\text{g S/mL}$  and diluting the mixture to 50.0 mL. The measured signal quotient was (signal due to X/signal due to S) = 1.690/1.000.
- (a) In a separate experiment in which the concentrations of X and S were equal, the quotient was (signal due to X/signal due to S) = 0.930/1.000. What is the concentration of X in the unknown? (5 pts)
- (b) Answer the same question if, in a separate experiment in which the concentration of X was 3.42 times the concentration of S, the quotient was (signal due to X/signal due to S) = 0.930/1.000. (5 pts)
- (3) State the advantages and disadvantages of a furnace compared a flame in atomic absorption spectroscopy. (5 pts)
- (4) Two solutes have a separation factor of  $\gamma = 1.06$ . How many plates are required to give a resolution of 1.0? of 2.0? If the plate height is 0.20 mm, how long must the column be for a resolution of 1.0? (5 pts)
- (5) The cell SCE||I<sup>-</sup> (x M), PbI<sub>2</sub> (s) | Pb
- (a) Develop an equation that relates the potential of the cell to pI. (5 pts)
- (b) Calculate pI if the cell has a potential of -0.348 V (5 pts)
- $$K_{sp} = [\text{Pb}^{2+}][\text{I}^{-}]^2 = 7.9 \times 10^{-9}$$
- $$E^{\circ}_{\text{Pb}^{2+}/\text{Pb}} = -0.126 \text{ V}$$
- $$E_{\text{SCE}} = 0.244 \text{ V}$$
- (6) A mixture of 14 compounds was subjected to a reversed-phase gradient separation going from 5% to 100% acetonitrile with a gradient time of 60 min. The sample was injected at  $t = \text{dwell time}$ . All peaks were eluted between 22 and 41 min.
- (a) Is the mixture more suitable for isocratic or gradient elution? (5 pts)
- (b) If the next run is a gradient, select the starting and ending % acetonitrile and the gradient time. (10 pts)