國立成功大學 112學年度碩士班招生考試試題

編 號:77

系 所: 化學工程學系

科 目: 有機化學

日 期: 0206

節 次:第1節

備 註:不可使用計算機

編號: 77

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考試日期:0206, 節次:1

第1頁,共4頁

※ 考生請注意:本試題不可使用計算機。 請於答案卷(卡)作答,於本試題紙上作答者,不予計分。

- I. Selective questions (4 points each, total 20 points)
 - [1] Which of the following bonds gives rise to a strong, broad, absorbance at approximately 3200-3650 cm⁻¹ in the infrared spectrum?

(a) C=O; (b) O-H; (c) C=C; (d) C-H

[2] Which of the following functional groups is likely to be present in a molecule that has peaks in the infrared spectrum at 3200-3650 cm⁻¹ (strong), but does not have peaks at 1630-1820 or 1000-1250 cm⁻¹

(a) alcohol; (b) aldehyde or ketone; (c) ether; (d) ester

[3] What type of mechanism accounts for the cleavage of dibutyl ether upon treatment with HBr to give two moles of 1-bromobutane?

(a) S_N1 ; (b) S_N2 ; (c) E1; (d) E2

[4] How many sets of equivalent protons are there in 3-methylhexane?

(a) 3; (b) 5; (c) 7; (d) 9

[5] How many stereoisomers of 3-bromo-2-butanol exist?

(a) 2; (b) 4; (c) 6; (d) 8

II. Provide the proper IUPAC name for the compound below. (5 points each, total 30 points)

[1]

[2]

系 所: 化學工程學系

考試科目:有機化學

考試日期:0206,節次:1

第2頁,共4頁

[3]

[4]

[5]

[6]

III. Draw the structure(s) (5 points each, total 45 points)

[1] Draw the structure of the major product in the following reaction.

$$\begin{array}{c|c} CH_3 \\ \hline \\ H_3O^+ \end{array} \begin{array}{c} Br_2 \\ \hline \\ FeBr_3 \end{array}$$

[2] Draw the structure of the major product in the following reaction.

系 所: 化學工程學系

考試科目:有機化學

考試日期:0206,節次:1

第3頁,共4頁

[3] Draw the structure of the major product in the following reaction:

[4] Draw two resonance structures for the species below.

[5] McMurry 5-19 Draw the structure of the major product in the following reaction:

[6] Draw the structure of the possible products in the following reaction:

[7] Draw the structure of the possible products in the following reaction and indicate the major product.

編號: 77

國立成功大學 112 學年度碩士班招生考試試題

系 所: 化學工程學系

考試科目:有機化學

考試日期:0206,節次:1

第4頁,共4頁

[8] Draw the structure of the major product in the following reaction:

[9] Brown 11-89. Draw the structure of the major product in the following reaction:

IV. Please rank the following acids in the order of decreasing acid strength (more acidic to less acidic)? FCH₂COOH, ClCH₂COOH, BrCH₂COOH (5 points)