

中原大學 100 學年度 碩士班 入學考試

3 月 19 日 13:30~15:00

物理學系、物理學系(在職)

誠實是我們珍視的美德，
我們喜愛「拒絕作弊，堅守正直」的你！

科目：量子物理

(共 2 頁第 1 頁)

可使用計算機，惟僅限不具可程式及多重記憶者

不可使用計算機

A. 單選題 (20 分，每題 5 分，答案必需依序填於答案卷)

1. The energy density of blackbody radiation is $u(\nu, T) = \frac{8\pi h}{c^3} \frac{\nu^3}{e^{h\nu/kT} - 1}$. At low frequency, this formula can be reduced to _____.

(A) $u(\nu, T) = -\frac{8\pi h \nu^3}{c^3}$ (B) $u(\nu, T) = \frac{8\pi \nu^2}{c^3} kT$

(C) $u(\nu, T) = \frac{8\pi h \nu^3}{c^3} e^{-h\nu/kT}$ (D) $u(\nu, T) = \frac{8\pi h^2 \nu^4}{c^3 kT}$

2. Consider a nonrelativistic electron whose energy is E . What is its de Broglie wavelength?

(A) $\frac{h}{E}$ (B) $\frac{E}{h}$ (C) $\frac{h}{\sqrt{2m_e E}}$ (D) $\frac{\sqrt{2m_e E}}{h}$

3. Consider the wavelength change $\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$ of the Compton scattering. The Compton wavelength of the electron is $\frac{h}{m_e c} = 2.426 \times 10^{-12}$ m. When the scattered photon has the maximum energy loss, what is the wavelength change?

(A) 0 (B) 1.213×10^{-12} m (C) 2.426×10^{-12} m (D) 4.852×10^{-12} m

4. The Hermite polynomials show up in the eigenfunctions of the harmonic oscillator. The polynomials can be expressed as $H_n(y) = (-1)^n e^{y^2} \frac{d^n}{dy^n} e^{-y^2}$. What is $H_3(y)$?

(A) $8y^3 - 12y$ (B) $8y^3 - 4y$ (C) $-8y^3 + 12y$ (D) $-8y^3 + 4y$

B. 計算題 (80 分，需寫出計算過程)

1. (20 points) The state of a particle is described by the following wave function:

$$\begin{aligned} \psi(x) &= A \quad \text{for } -a < x < 2a \\ &= 0 \quad \text{for } x < -a \text{ and } x > 2a \end{aligned}$$

(a) A is real. Find A by using the normalization condition.

(b) What is the probability of finding the particle between $x = 0$ and $x = a$?

(c) Calculate $\langle x \rangle$ and $\langle x^2 \rangle$ for this state.

(d) Calculate the momentum probability density.

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2. (20 points) Consider a simple harmonic oscillator system with Hamiltonian $H = \frac{p^2}{2m} + \frac{m\omega^2 x^2}{2}$.

Define the operators A and A^+ as: $A = \sqrt{\frac{m\omega}{2\hbar}}x + i\frac{p}{\sqrt{2m\omega\hbar}}$ and $A^+ = \sqrt{\frac{m\omega}{2\hbar}}x - i\frac{p}{\sqrt{2m\omega\hbar}}$.

(a) Find the commutation relations $[H, A]$, $[H, A^+]$, and $[A, A^+]$.

(b) When a perturbation λH_1 is introduced, $H \rightarrow H + \lambda H_1$. We can expect that the energy

eigenvalues $E_n \rightarrow E_n + \lambda E_n^{(1)} + \lambda^2 E_n^{(2)} \dots$. Suppose the perturbation is $\lambda H_1 = \lambda x$. Calculate the

first-order energy shift $\lambda E_n^{(1)}$ and the second-order energy shift $\lambda^2 E_n^{(2)}$.

3. (20 points) (a) Use the Schrödinger equation to calculate the eigenfunctions and the energy eigenvalues for an electron in a three-dimensional cubical box with side L .

(b) Consider the Pauli exclusion principle. What is the lowest energy of a set of 25 electrons in a cubical box with side L ?

4. (10 points) Consider the addition of spins for a two-electron system. We can denote the spinor of the first electron by $\chi_{\pm}^{(1)}$, and similarly for the spinor $\chi_{\pm}^{(2)}$ of the second electron. Use the combinations of $\chi_{\pm}^{(1)}$ and $\chi_{\pm}^{(2)}$ to represent the triplet states and the singlet state.

5. (10 points) An electron in a hydrogen atom is in a state described by the wave function

$$\psi(\vec{r}) = A(\psi_{100}(\vec{r}) - 2\psi_{200}(\vec{r}) + 3\psi_{211}(\vec{r}) - \sqrt{2}\psi_{210}(\vec{r}))$$

(a) Find A by using the normalization condition.

(b) What are the expectation values of L_z and \vec{L}^2 ?