

※ 考生請注意：本試題可使用計算機。請於答案卷(卡)作答，於本試題紙上作答者，不予計分。

1. A mass-and atmosphere-loaded piston/cylinder contains 2 kg of water at 50°C, 200 kPa. Heat is added from a reservoir at 200°C to the water until it reaches 200°C. Find the work, heat transfer, and total entropy production for the system and surroundings. (15%)

2. A mass-and atmosphere-loaded piston/cylinder initially contains 0.5 kg of steam at 1.2 MPa, superheated by 112°C. Now, steam loses heat to the surroundings and the piston moves down hitting a set of stops at which point the cylinder contains saturated liquid water. The cooling continues until the cylinder contains water at 100°C. Determine
 - (a) the initial temperature. (3%)
 - (b) the specific volume of the steam by the time the piston first hits the stops. (3%)
 - (c) the final pressure and the quality (if mixture). (6%)
 - (d) the total heat transfer. (5%)
 - (e) show the processes in a P-v diagram. Please draw the wet(vapor) dome and clearly mark each state. (3%)

3. A mass-and atmosphere-loaded piston/cylinder assembly contains 1 kg of water at 25°C, 200 kPa. A nozzle in a line to the cylinder is opened to enable a flow to the outside atmosphere at 100 kPa. The process continues until half the mass has flowed out. Assume there is no heat transfer and water temperature is constant. Find the exit velocity, total work done, and total entropy generation in the process. (15%)

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Property tables for problems 1, 2 and 3. If there is no appropriate table, please implement suitable approximation.

Saturated Water

T (°C)	P (kPa)	v (m ³ /kg)		u (kJ/kg)		h (kJ/kg)		s (kJ/kg-K)	
		v_f	v_g	u_f	u_g	h_f	h_g	s_f	s_g
25	3.169	0.001003	43.3593	104.86	2409.76	104.87	2547.17	0.3673	8.5579
50	12.350	0.001012	12.0318	209.30	2443.47	209.31	2592.06	0.7037	8.0762
75	38.58	0.001026	4.13123	313.87	2475.91	313.91	2635.28	1.0514	7.6824
100	101.3	0.001044	1.67290	418.91	2506.50	419.02	2676.05	1.3068	7.3548
188	1200	0.001139	0.16333	797.27	2588.82	798.64	2784.82	2.2165	6.5233

Superheated Vapor Water

T (°C)	P (kPa)	v (m ³ /kg)	u (kJ/kg)	h (kJ/kg)	s (kJ/kg-K)
200	200	1.08034	2654.39	2870.46	7.5066
200	1200	0.16930	2612.74	2815.90	6.5898
250	1200	0.19235	2704.20	2935.01	6.8293
300	1200	0.2138	2789.22	3045.80	7.0316
350	1200	0.23452	2872.16	3153.59	7.2120

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4. (12%) Derive an expression for the enthalpy change in an isothermal process of a gas for which the equation of state is $P = \frac{RT}{(v-a)} - \frac{1}{v^2}$, where a is constant, and P , v , T , R are pressure, specific volume, temperature and gas constant, respectively.

5. The compression ratio in an air-standard Otto cycle is 10. At the beginning of the compression stroke, the pressure is 0.1 MPa and the temperature is 25 °C. The heat transfer to the air per cycle is 2000 kJ/kg air. Use 1.4 for specific heat ratio, 0.287 kJ/kg-K for gas constant and 0.717 kJ/kg-K for constant volume specific heat. Determine

- (1) (8%) Sketch P - v and T - s diagrams of the cycle.
- (2) (18%) The pressure and temperature at the end of each process of the cycle.
- (3) (6%) The thermal efficiency.
- (4) (6%) The mean effective pressure.