

國立雲林科技大學 108 學年度

碩士班招生考試試題

系所:化材系

科目: 有機化學

1. Consider the equilibrium

$$PO_4^{3-} + H_2O$$

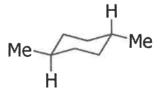
$$HPO_4^{2-} + HO^{-}$$

Which are the Brönsted-Lowry bases? (6%, more than one answer, give scores depending on the degree of correctness)

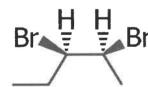
2. (a) Calculate the index of hydrogen deficiency (or degree of unsaturation) of the following molecule. (5%)

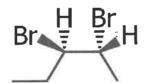
- (b) Calculate the index of hydrogen deficiency (or degree of unsaturation) of a compound with the molecular formula of C7H7NOS. (5%)
- 3. Label the following pairs of molecules as enantiomers, diastereomers, or same molecule. (9%)

(a)

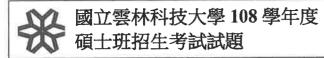


(b)





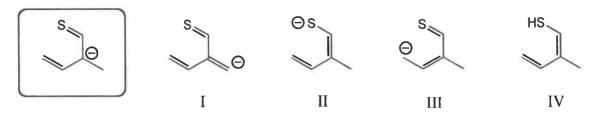
(c)



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4. Which of the following species are *not* a resonance form(s) of the anionic species in the box? (6%, more than one answer, give scores depending on the degree of correctness)



5. Which will be the major products of the following reactions? (9%)

(a)

(b)

(c)

6. Which of the following represent (R)-2-butanol? (10%, more than one answer, give scores depending on the degree of correctness)

$$H_3C$$
 H_3 H_3C H_3 H_4 H_5 H_5 H_5 H_5 H_5 H_6 H_7 H_8 $H_$



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7. Provide the reagents for transformations (1), (2), and (3). (15%)

$$(1) \longrightarrow (2) \longrightarrow (3) \longrightarrow (3)$$

- 8. When benzene reacts with 1-chloro-2,2-dimethylpropane (neopentyl chloride) in the presence of aluminum chloride, the major product is 2-methyl-2-phenylbutane, not 2,2-dimethyl-1-phenylpropane (neopentylbenzene). Explain this result. (10%)
- Rank the following dienes in order of increasing reactivity in a Diels-Alder reaction (1 = least reactive, 4 = most reactive). Briefly explain your ranking. (5%)

10. Provide mechanistic explanations for the following observations: (10%)

(a)
$$\xrightarrow{HCI}$$
 \xrightarrow{CI} \xrightarrow{CI} \xrightarrow{CI} \xrightarrow{CI} \xrightarrow{CI}



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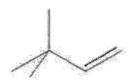
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11. Starting with an appropriate alkyl halide and base, outline syntheses that would yield each of the following alkenes as the major (or only) product: (10%)

(a)



(b)

