題號: 93 國立臺灣大學108學年度碩士班招生考試試題

科目:分析化學(A)

節次: 6

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1. Complete the following reactions (20%)

(i)
$$V^{2+} + Fe^{3+} + H_2O \rightarrow$$

(ii)
$$Cr_2O_7^{2-} + Fe^{2+} + H^+ \rightarrow$$

(iii)
$$I_3^- + H_2S_{(g)} \rightarrow$$

(iv)
$$BrO_3 + Br + H^+ \rightarrow$$

- 2. Carbon dioxide is important to world climate. Describe how is gaseous carbon dioxide dissolved in ocean, and dissociated into different dissolved species (10%). Assume the first dissociation constant K_1 is 10^{-6} and the second dissociation constant K_2 is 10^{-10} , draw a diagram to show the fractionation of different species against pH (10%). At pH=8, what is the percentage of HCO_3^- and CO_3^{-2} ? (10%)
- 3. Give a general description of the principle of chromatography. (10%) What is HETP? (5%) Write out the van Deemter Equation, (5%) draw a diagram to show the relationship of each term against flow speed u. (5%) Derive the optimal flow speed $u_{\rm opt}$. (5%)
- 4. Describe the difference between "precision" and "accuracy". (5%) What are: Coefficient of variation? Standard deviation? Relative standard deviation? 95% of confidence level? Q-test? (15%)

試題隨卷繳回