

1. (30 %, 15% for each) Van der Waals gases obey the following equation of state (van der Waals equation):

$$P = \frac{RT}{\bar{V} - b} - \frac{a}{\bar{V}^2}$$

Where a , and b are coefficients, R is gas constant ($8.3145 \text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$), \bar{V} is molar volume ($\frac{V}{n}$), T is temperature in unit of K and P is pressure.

(a) $\left(\frac{\partial U}{\partial V}\right)_T$ indicates how internal energy changes with the change of volume under constant temperature and it obeys the equation shown below:

$$\left(\frac{\partial U}{\partial V}\right)_T = T \left(\frac{\partial P}{\partial T}\right)_V - P$$

where P is pressure, T is temperature and V is volume. Show that $\left(\frac{\partial U}{\partial V}\right)_T = \frac{an^2}{V^2}$ for van der Waals gases.

(b) The change of the internal energy U could be represented as following:

$$dU = \left(\frac{\partial U}{\partial T}\right)_V dT + \left(\frac{\partial U}{\partial V}\right)_T dV$$

Calculate ΔU_m for the isothermal reversible expansion of 1 mole argon (Van der Waals coefficients :

$a = 1.337 \text{ atm L}^2 \text{ mole}^{-2}$; $b = 3.20 \times 10^{-2} \text{ L mole}^{-1}$) for an initial volume of 1.00 L to 24.0 L at constant temperature of 298K.

2. (20%) The molar heat capacities $C_{p,m}$ (unit: $\text{J} \cdot \text{K}^{-1} \cdot \text{mole}^{-1}$) of N_2 at constant pressure obeys the following relation with temperature: $C_{p,m} = a + bT + \frac{c}{T^2}$ where T is temperature

and $a = 28.58 \text{ J} \cdot \text{K}^{-1} \cdot \text{mole}^{-1}$, $b = 3.77 \times 10^{-3} \text{ J} \cdot \text{K}^{-2} \cdot \text{mole}^{-1}$ and $c = -0.5 \times 10^5 \text{ J} \cdot \text{K} \cdot \text{mole}^{-1}$

The change of molar entropy due to the change of temperature under constant pressure is as following :

$$\Delta S_m = \int_{T_i}^{T_f} \frac{C_{p,m}}{T} dT$$

What is the change of molar entropy of N_2 when it is heat from 298K to 373K.

國立交通大學 107 學年度碩士班考試入學試題

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系所班別：生物科技學系 組別：生科系丙組

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【不可使用計算機】*作答前請先核對試題、答案卷(試卷)與准考證之所組別與考科是否相符!!

3. To calculate

(a) (5%) $\lim_{x \rightarrow 0} \frac{2 \tan^2 x}{1 - \cos^2 x}$

(b) (5%) $\lim_{n \rightarrow \infty} (\sqrt{n^2 + n} - n)$

(c) (5%) $\lim_{x \rightarrow +\infty} \frac{\ln(\ln x)}{\ln(x - \ln x)}$

4. To calculate

(a) (5%) $\int_0^{3\pi} |\sin x| dx$

(b) (5%) $\int e^{-\sqrt{x}} dx$

(c) (5%) $\int_0^\pi \int_x^\pi \frac{\sin y}{y} dy dx$

(d) (5%) $\int_{-3}^3 \int_{-\sqrt{9-x^2}}^{\sqrt{9-x^2}} \sqrt{9-x^2-y^2} dy dx$

5. To calculate the following statements.

(a) (5%) Determine the area of the region enclosed by $y = 2x^2$ and $y = -3x^2 + 5$

(b) (5%) Let A be the area found in question a. Determine the volume of the solid obtained by rotating area A around x-axis

(c) (5%) Determine the volume of the solid obtained by rotating area A around y-axis