編號: 318

國立成功大學 106 學年度碩士班招生考試試題

所:環境醫學研究所

考試科目:環境化學

考試日期:0214, 節次:3

第1頁,共1頁

※ 考生請注意:本試題不可使用計算機。 請於答案卷(卡)作答,於本試題紙上作答者,不予計分。

- 1. Balance the following equations: (25%, 5% for each)
 - (1) $CaCl_2 + Na_2CO_3 \rightarrow CaCO_2 + NaCl$
 - (2) $Cl_2 + KOH \rightarrow KClO_3 + KCl + H_2O$
 - (3) $C_6H_{12}O_6 + NO_3 + H^+ \rightarrow H_2O + N_2 + 6 CO_2$
 - (4) $CH_3COO^{-} + Cr_2O_7^{-2} + H^{+} \rightarrow CO_2 + N_2 + H_2O$ (5) $SO_3^{-2} + Fe^{3+} + H_2O \rightarrow SO_4^{-2} + Fe^{2+} + H^{+}$
- 2. Please use chemical reaction equations to explain how the NOx and CFC to destroy the stratospheric ozone layer respectively? (15%)
- 3. Chlorination is the major disinfection method of tap water, please use figure and reactive equation to interpret what are (1) Breakpoint Chlorination? (2) Free and Combined Chlorine Residuals? (10%) In addition, please present two analytical methods to measure the Free and Combined Chlorine Residuals in water samples.(10%)
- 4. Please describe the formation mechanism of secondary PM2.5? (20%)
- 5. What is the function of the NaOH sometimes used in preparing the thiosulfate solution used for dissolved oxygen measurement? (10%)
- 6. What is the difference between orthophosphates, polyphosphates, and organic phosphorus? (6%) In which form must the phosphorus be for colorimetric analysis? (4%)