

國立中山大學 104 學年度碩士暨碩士專班招生考試試題

科目名稱：有機化學【海資系碩士班丙組】

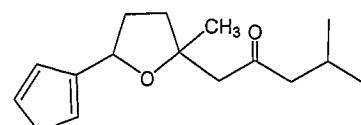
題號：452001

※本科目依簡章規定「不可以」使用計算機(選擇題)

共 3 頁第 1 頁

Choose one right answer from a list of answers given for each question. (5 points for each)

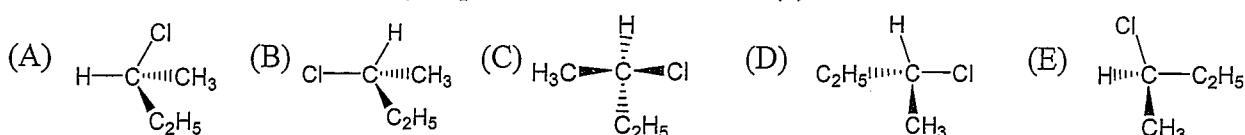
1. For the compound



, please show how many isoprene units are

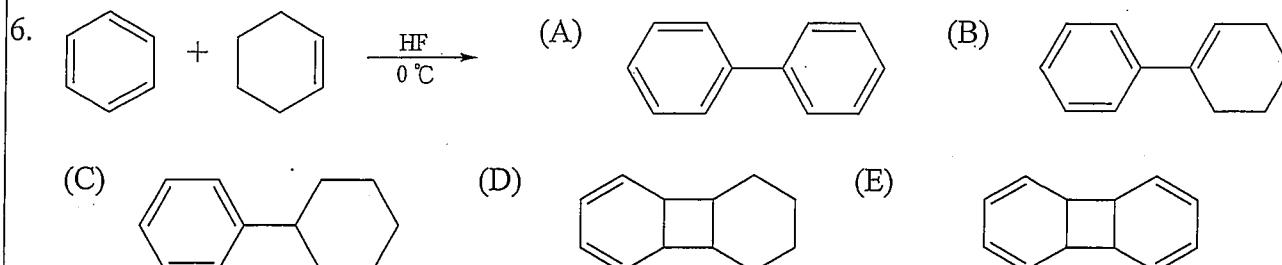
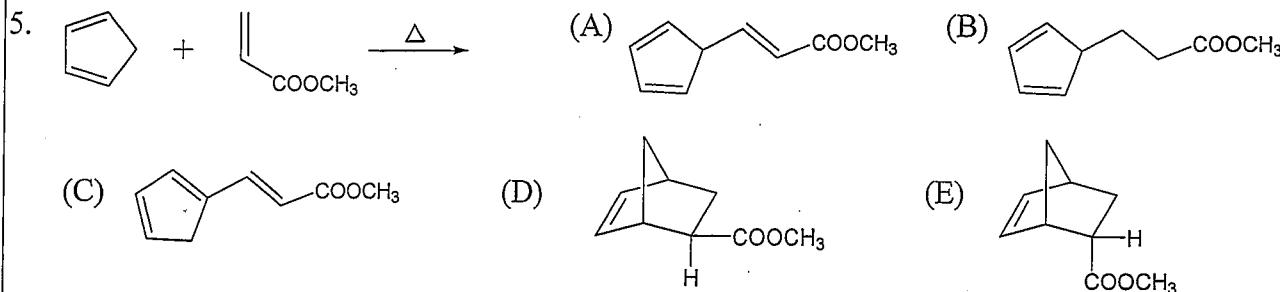
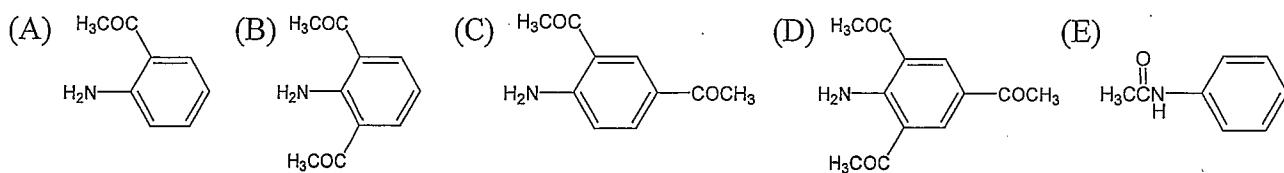
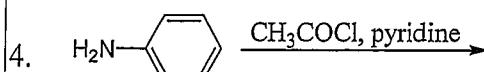
incorporated into the structure? (A) 2 (B) 3 (C) 4 (D) 5 (E) 6

2. Which of the following drawings represents the structure of (*S*)-2-chlorobutane?



3. How many possible aldol condensation products could be obtained in a 1 : 1 mixture of acetaldehyde and propanal? (A) 8 (B) 6 (C) 4 (D) 2 (E) 1

Show the product (or major product) for each of the following reactions (problems 4–10).



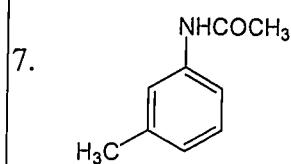
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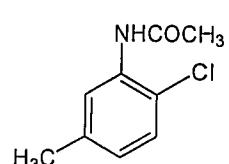
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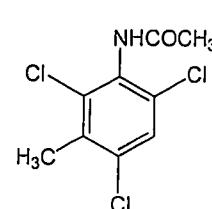
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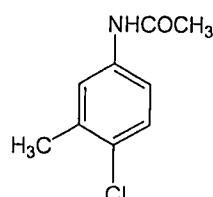
(A)



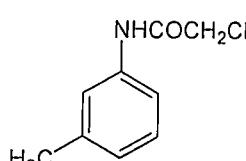
(B)



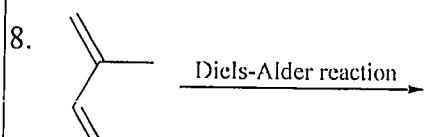
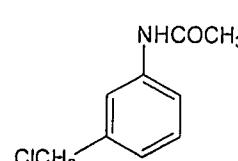
(C)



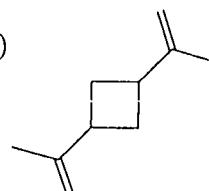
(D)



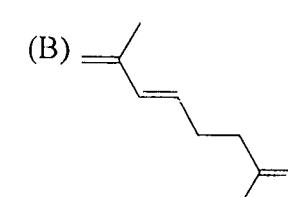
(E)



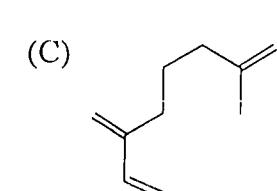
(A)



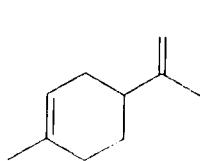
(B)



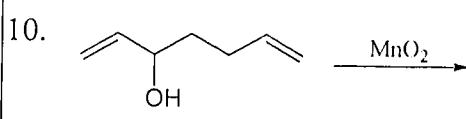
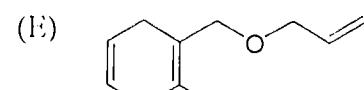
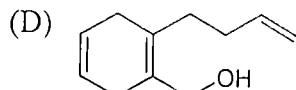
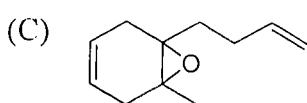
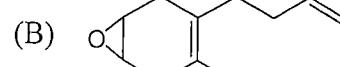
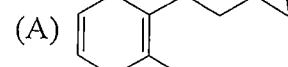
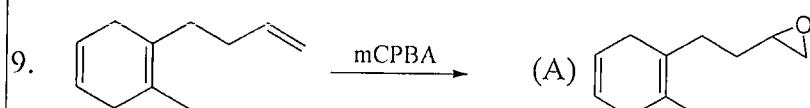
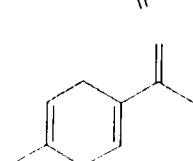
(C)



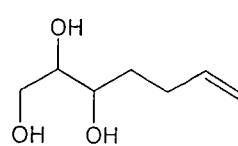
(D)



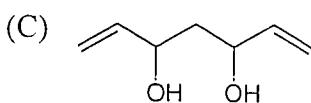
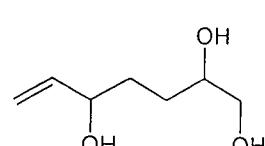
(E)



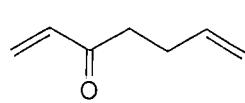
(A)



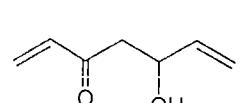
(B)



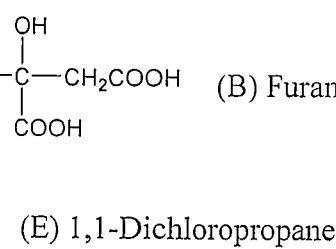
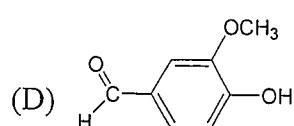
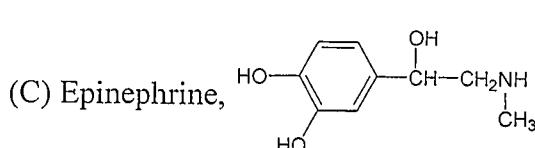
(D)



(E)



11. Which of the following compounds is chiral? (A) Citric acid, (B) Furan



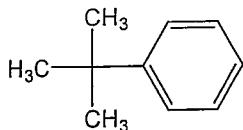
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12.



$\xrightarrow{\text{SO}_3, \text{conc. H}_2\text{SO}_4}$

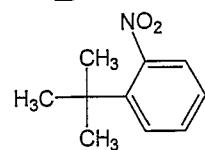
A

$\xrightarrow{\text{HNO}_3, \text{H}_2\text{SO}_4}$

B

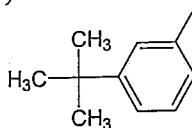
$\xrightarrow[\Delta]{\text{H}^+, \text{H}_2\text{O}}$

C

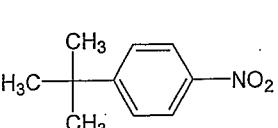


What is the final product C of the above reaction sequence? (A)

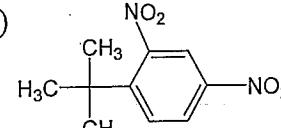
(B)



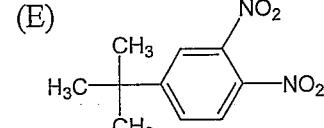
(C)



(D)



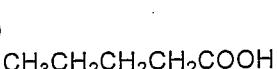
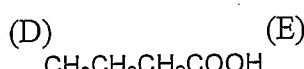
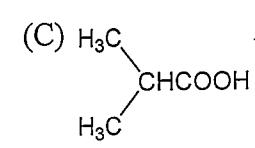
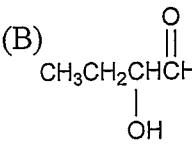
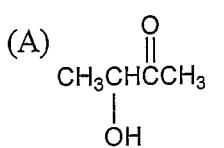
(E)



13. Show the degrees of unsaturation (or index of hydrogen deficiency) for a compound with the molecular formula $\text{C}_{21}\text{H}_{21}\text{BrN}_2\text{O}_2$. (A) 9 (B) 10 (C) 11 (D) 12 (E) 13

14. A compound C_5H_{10} shows IR absorptions at 3030, 1660 (very weak), 1380 and 965 cm^{-1} . This compound is (A) 1-pentene (B) cyclopentane (C) cis-2-pentene (D) 2-methyl-2-butene (E) trans-2-pentene.

15. Which of the following compounds could show a molecular ion peak at $m/z = 88$ and a base peak at $m/z = 60$ in EI mass spectrum?

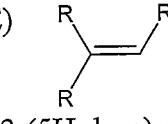
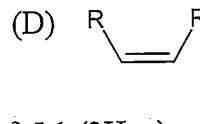
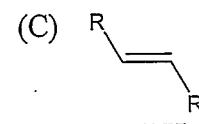
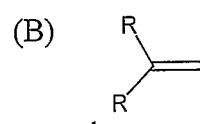
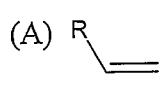


16. The mass spectrum of a compound showed a relative intensity ratio 9 : 6 : 1 for peaks of M (molecular ion), M+2 and M+4. This compound should possess (A) Br_2 (B) Cl_2 (C) BrCl (D) Br_2Cl (E) BrCl_2 in the molecular formula.

17. The ^{13}C NMR spectrum of a compound with the molecular formula C_6H_{14} showed four peaks at δ 9, 29, 30 and 37 ppm, respectively. The peak at 29 ppm is much larger than the others, whereas the peak at 30 ppm is very weak. What is this compound? (A) 2,3-Dimethylbutane (B) 2,2-Dimethylbutane (C) 3-Methylpentane (D) n-Hexane (E) 2-Methylpentane

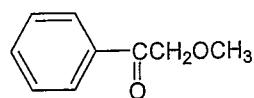
18. The ^1H NMR spectrum of a compound $\text{C}_{10}\text{H}_{20}$ shows two peaks at δ 0.97 (18H, s) and δ 5.30 (2H, s) ppm. How many carbon signals of this compound could be found in its ^{13}C NMR spectrum? (A) 2 (B) 3 (C) 4 (D) 5 (E) 6

19. Which of the following compounds would you expect to have IR absorptions at 910 and 990 cm^{-1} ?

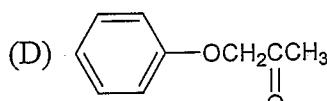
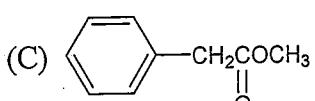
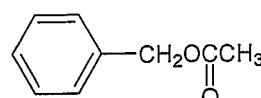


20. A compound showed ^1H NMR signals at δ 2.0 (3H, s), δ 5.1 (2H, s) and δ 7.3 (5H, br s) ppm. The

structure of this compound is (A)



(B)



(E)