## 淡江大學 101 學年度碩士班招生考試試題

系別:化學學系 科目:無機化

考試日期:2月26日(星期日) 第3節

本試題共 五 大題, 壹 頁

## 1. 按照題目順序作答. 2. 每題 20 分. 3. 需寫出詳細的推理過程,不可只有答案.

- 1. a) Predict the electronic spectrum of (NH<sub>4</sub>)V(SO<sub>4</sub>)<sub>2</sub>·12H<sub>2</sub>O in an aqueous solution. Explain these observations. b) Predict the spin-only magnetic moment of this complex.
- 2. Show diagrammatically the splitting of **d** orbital, energy-wise, in (a) square pyramidal (b) tetrahedral and (c) square planar fields. Would the following species to be expected to be a Jahn-Teller distorted structure?(d) [CoF<sub>6</sub>]<sup>3</sup>-(e) [Cu(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup>
- 3. Describe the bonding and characteristics of a) Fischer carbene and b) Schrock carbene. Give a real example and explanation of the following reactions: c)  $\alpha$  abstraction d)  $\beta$  elimination e) methyl migration f) oxidative addition.
- Assign the point group for the following molecules. (a) ferrocene (b) B(OH)<sub>3</sub>
  (c) cis-[FeCl<sub>2</sub>(H<sub>2</sub>O)<sub>4</sub>]. d) Using the group theory give an MO description of bonding in NH<sub>3</sub> and draw qualitative energy level diagram.
- 5. Give Lewis structures and sketch the shapes of the following.
  - a)  $IF_3$  b)  $NO_2$ + c)  $SF_4$ .
  - d) What intramolecular and intermolecular chemical forces exist in each species: H<sub>2</sub>O, CaO, SiO<sub>2</sub>, CO<sub>2</sub>? e) List the following in order of increasing boiling point: H<sub>2</sub>O, Xe, LiO<sub>2</sub>, CaO, SiO<sub>2</sub>, CO<sub>2</sub>.

C <sub>3v</sub>	E	$2C_3$	$3\sigma_v$	
$A_1$	1	1	1	Z
$A_2$	1	1	-1	
E	2	-1	0	(x,y)