

國立臺灣師範大學 103 學年度碩士班招生考試試題

科目：無機化學

適用系所：化學系

注意：1.本試題共 2 頁，請依序在答案卷上作答，並標明題號，不必抄題。2.答案必須寫在指定作答區內，否則不予計分。

- (1) Draw the crystal field splitting energy diagram of the d-block complexes that have a trigonal Bipyramidal geometry. (10 points)
- (2) A tetracoordinated complex $Mabcd$ is known to have stereoisomers. Please draw the 3-D structures for all the possible stereoisomers. (M stands for the central metal ion and a, b, c, and d each stand for a monodentate ligand) (10 points)
- (3) Draw and mark the symbols (Δ and Λ) for the enantiomers of $Cr(acac)_3$. (10 points)
- (4) The magnetic moment of $[Mn(en)_2Br(NO)]^+$ is $6.06 \mu_B$ (B.M.) and en is the abbreviation for ethylenediamine. Please answer the following questions. (20 points)
 - (a) Draw the structure for en.
 - (b) How many lone pair electrons does the central Mn ion have?
 - (c) Draw the d-orbital energy splitting diagram for the Mn ion.
 - (d) Draw all stereoisomers for $[Mn(en)_2Br(NO)]^+$.
- (5) Calculate the lattice energy for $NaCl_{(s)}$ ($Na^+_{(g)} + Cl^-_{(g)} \rightarrow NaCl_{(s)}$) by Born-Haber cycle. Ionization energy of $Na_{(g)}$ is 496 kJ/mol, sublimation of $Na_{(s)}$ is 108 kJ/mol, dissociation of $Cl_{2(g)}$ is 244 kJ/mol, electron affinity of $Cl_{(g)}$ is 349 kJ/mol, and formation of $NaCl_{(s)}$ is -411 kJ/mol. (10 points)
- (6) Which of the following species are nonpolar? (a) PH_3 (b) H_2Se (c) CH_2Cl_2 (d) IF_5 (e) NO_3^- (f) O_3 (g) SO_4^{2-} (h) NH_4^+ (i) I_3^- (j) PCl_5 (10 points)
- (7) Construct the MO diagram of N_2 and use the diagram to explain why N_2 is a very stable molecule. (10 points)

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- (8) What is the ratio of the edge length of the unit cell for a body-centered cubic crystal to the radius of the atoms in the crystal? (10 points)
- (9) The point group of $\text{Mn}(\text{CO})_5\text{Cl}$ is C_{4v} . Please find the IR active modes for the CO stretching vibrations of $\text{Mn}(\text{CO})_5\text{Cl}$? (10 points)

C_{4v}	E	$2C_4$	C_2	$2\sigma_v$	$2\sigma_d$		
A_1	1	1	1	1	1	z	$x^2 + y^2, z^2$
A_2	1	1	1	-1	-1	R_z	
B_1	1	-1	1	1	-1		$x^2 - y^2$
B_2	1	-1	1	-1	1		xy
E	2	0	-2	0	0	$(x, y)(R_x, R_y)$	(xz, yz)